



**UNIVERSITI TUN HUSSEIN ONN MALAYSIA**

**FINAL EXAMINATION  
SEMESTER II  
SESSION 2012/13**

**COURSE NAME** : ENVIRONMENTAL SAMPLING  
**COURSE CODE** : DAC 38003  
**PROGRAMME** : 3 DFT  
**EXAMINATION DATE** : MARCH 2013  
**DURATION** : 3 HOURS  
**INSTRUCTION** : ANSWER 5 (FIVE) QUESTIONS ONLY

THIS QUESTION PAPER CONSISTS OF FOUR (4) PAGES

- Q1**
- (a) List 3 (**THREE**) water quality parameters. (3 marks)
- (b) Describe the method of preparing iodine solution. (5 marks)
- (c) List 3 (**THREE**) applications of iodine solution. (3 marks)
- (d) Explain briefly the method of waste disposal for the following: (9 marks)
- i) On-site recovery
  - ii) On-site storage
  - iii) Prescribed premises
- Q2**
- (a) Briefly explain sample distillation. (4 marks)
- (b) Describe the differences between settleable solids and nonsettleable solids. (4 marks)
- (c) Explain the application of extraction dilution for wastewater sample. (4 marks)
- (d) Sketch and label the Soxhlet Extractor. (8 marks)

- Q3** (a) List **5 (FIVE)** steps in selecting sample container. (5 marks)
- (b) Describe the preparation of experimental apparatus. (5 marks)
- (c) Explain briefly the sterillization method for the following: (10 marks)
- i) Derivation of hydrochloric acid
  - ii) UV light
  - iii) Filter paper
  - iv) Radiation
  - v) Sterillization efficiency test
- Q4** (a) Describe the preservation method of the inorganic chemical reagent. (4 marks)
- (b) Explain the reason for instrumental calibration. (4 marks)
- (c) Explain the procedures of calibration activities. (12 marks)

- Q5** (a) What is bioconcentration?  
(6 marks)
- (b) Describe the determination of bioconcentration factor.  
(4 marks)
- (c) Determine approximate alkalinity at 25°C for a water containing 100 mg/L  $\text{CO}_3^{2-}$  and 75 mg/L  $\text{HCO}_3^-$  at pH 10.  
(C = 12, O = 16, H = 1,  $\text{H}^+ = 10^{-7}$  and  $\text{OH}^- = 1.7$ )  
(10 marks)
- Q6** (a) List 5 (FIVE) safety rules of the in-situ test.  
(10 marks)
- (b) Determine the  $\text{BOD}_5$  for the 10 mL of wastewater and 300 mL of dilution water filled up in BOD bottle. Concentration of dissolved oxygen for day-1 is 7.5 mg/L and for day-5 is 5.0 mg/L.  
(5 marks)
- (c) Determine the fraction of maximum oxygen consumption in four days for a wastewater being discharged into a river at temperature 10°C and BOD rate constant under standard conditions is  $0.115 \text{ day}^{-1}$  with  $\theta$  value is 1.135.  
(5 marks)

- END OF QUESTION -