

CONFIDENTIAL



UNIVERSITI TUN HUSSEIN ONN MALAYSIA

**FINAL EXAMINATION  
SEMESTER II  
SESSION 2016/2017**

COURSE NAME : ENGINEERING TECHNOLOGY MATERIALS  
COURSE CODE : BDU 10603  
PROGRAMME CODE : 1 BDC /1 BDM  
EXAMINATION DATE : JUNE 2017  
DURATION : 3 HOURS  
INSTRUCTION : ANSWERS **FIVE (5)** QUESTIONS **ONLY**

THIS QUESTION PAPER CONSISTS OF **SIX (6)** PAGES

WATAUBUN 75 SHARIP...  
ALBYA JANI MHD...  
TANIR BTNS...  
MS'CT JURAU BOHCT

**TERBUKA**

**CONFIDENTIAL**

- Q1** (a) List four (4) mechanical properties of metals. (4 marks)
- (b) In cubic unit cells, sketch the following direction vectors and crystal planes
- (i)  $[0\bar{1}2]$
  - (ii)  $[1\bar{2}1]$
  - (iii)  $(\bar{1}2\bar{3})$
  - (iv)  $(0\bar{1}\bar{3})$
- (8 marks)
- (c) Calculate the atomic radius of Vanadium, given that Vanadium has a BCC structure, a density of  $5.96\text{g/cm}^3$  and an atomic weight of  $50.9\text{ g/mol}$  (8 marks)
- Q2** (a) Compare the mechanism of interstitial and vacancy atomic for diffusion. (4 marks)
- (b) Briefly explain two (2) reasons why interstitial diffusion is normally more rapid than vacancy diffusion. (4 marks)
- (c) An FCC iron-carbon alloy initially containing 0.35 wt% C is exposed to an oxygen-rich and virtually carbon-free atmosphere at 1400 K ( $1127^\circ\text{C}$ ). Under these circumstances the carbon diffuses from the alloy and reacts at the surface with the oxygen in the atmosphere; that is, the carbon concentration at the surface position is maintained essentially at 0 wt% C. This process of carbon depletion is termed decarburization. At what position will the carbon concentration be 0.15 wt% after a 10-h treatment? The value of D at 1400 K is  $6.9 \times 10^{-11}\text{ m}^2/\text{s}$ . (The Error Function value is given in Table Q2 (b)). (8 marks)
- (d) The activation energy for the diffusion of carbon in chromium is 111,000 J/mol. Calculate the diffusion coefficient at 1100 K ( $827^\circ\text{C}$ ), given that D at 1400 K ( $1127^\circ\text{C}$ ) is  $6.25 \times 10^{-11}\text{ m}^2/\text{s}$ . (4 marks)
- Q3** (a) A specimen of ductile cast iron having a rectangular cross section of dimensions 4.8 mm  $\times$  15.9 mm is deformed in tension and the results of the tensile test are given in Table Q3(a). Plot the engineering stress-strain curve for these data. (Note that 1 MPa = 1 N/mm<sup>2</sup>)

(8 marks)

- (b) From the engineering stress strain curve plotted in Q3(a), determine :
- (i) The 0.2% offset yield strength
  - (ii) The modulus of elasticity
  - (iii) The ultimate tensile strength
  - (iv) The percent of elongation

( 8 marks)

- (c) Gives four (4) methods to improve fatigue resistance of structural components.

(4 marks)

- Q4** (a) Consider the Copper-Lead (Cu-Pb) phase diagram system in Figure Q4(a). Make the phase analyses which includes of the phases present, the chemical compositions of the phases and the amounts of each phase of Cu-10 wt % Pb at the temperatures of :
- (i) 1000°C
  - (ii) 955°C +  $\Delta T$ ,
  - (iii) 955°C- $\Delta T$ ,
  - (iv) 200°C.

(16 marks)

- (b) Briefly explain the differences between pearlite, bainite, and spheroidite relative to microstructure and mechanical properties.

(4 marks)

- Q5** (a) Describe the properties and gives two (2) examples for each of these metals:

- (i) Ferrous metals
- (ii) Non-ferrous metals

(8 marks)

- (b) Suggest three (3) advantages and disadvantages of using advanced ceramics in automobile engines application.

(6 marks)

- (c) Describe and illustrate four (4) types of Polymer microstructure. In the illustration, arrange the microstructure based on their covalent strength.

(6 marks)

- Q6** (a) Continuous glass-fiber-reinforced-epoxy composite is made by 60% volume of E-glass fibers having a modulus of elasticity of  $7.24 \times 10^4$  MPa and a tensile strength of 2.4 GPa; and a hardened epoxy resin with a modulus of  $3.1 \times 10^3$  MPa and a tensile strength of 0.06 GPa. Assume the stress on the material causes uniform strain on all the composite layers. Calculate:
- (i) The modulus of elasticity of this composite
  - (ii) The tensile strength
- (6 marks)
- (b) Briefly explains the benefits of using Polymer Matrix Composite (PMC), Metal Matrix Composites (MMC) and Ceramic Matrix Composites (CMC) with regards to the toughness, modulus of Elasticity and creep resistance of the materials, by providing relevance graph.
- (6 marks)
- (c) Hand layup and spray up techniques, Filament winding, Pultrusion, Autoclave moulding and Resin Transfer Moulding (RTM) are types of forming processes for Thermosets matrix composites. Choose two (2) of this forming process, explain the process and evaluate the advantages or disadvantages of the chosen process.
- (8 marks)

-END OF QUESTIONS –

**FINAL EXAMINATION**

SEMESTER / SESSION : SEM II / 2016/2017  
 COURSE NAME : ENGINEERING TECHNOLOGY MATERIALS

PROGRAMME CODE: 1 BDC/ 1 BDM  
 COURSE CODE : BDU 10603

**Table Q2(b) : Tabulation of Error Function Values**

<i>z</i>	<i>erf(z)</i>	<i>z</i>	<i>erf(z)</i>	<i>z</i>	<i>erf(z)</i>
0	0	0.55	0.5633	1.3	0.9340
0.025	0.0282	0.60	0.6039	1.4	0.9523
0.05	0.0564	0.65	0.6420	1.5	0.9661
0.10	0.1125	0.70	0.6778	1.6	0.9763
0.15	0.1680	0.75	0.7112	1.7	0.9838
0.20	0.2227	0.80	0.7421	1.8	0.9891
0.25	0.2763	0.85	0.7707	1.9	0.9928
0.30	0.3286	0.90	0.7970	2.0	0.9953
0.35	0.3794	0.95	0.8209	2.2	0.9981
0.40	0.4284	1.0	0.8427	2.4	0.9993
0.45	0.4755	1.1	0.8802	2.6	0.9998
0.50	0.5205	1.2	0.9103	2.8	0.9999

**Table Q3(a): The tensile test of a Cast Iron**

Load		Length	
N	lbf	mm	in
0	0	75	2.953
4,740	1,065	75.025	2.954
9,140	2,055	75.05	2.955
12,920	2,900	75.075	2.956
16,540	3,720	75.113	2.957
18,300	4,110	75.15	2.959
20,170	4,530	75.225	2.962
22,900	5,145	75.375	2.968
25,070	5,635	75.525	2.973
26,800	6,025	75.75	2.982
28,640	6,440	76.5	3.012
30,240	6,800	78	3.071
31,100	7,000	79.5	3.13
31,280	7,030	81	3.189
30,820	6,930	82.5	3.248
29,180	6,560	84	3.307
27,190	6,110	85.5	3.366
24,140	5,430	87	3.425
18,970	4,265	88.725	3.493

Fracture

**TERBUKA**

FINAL EXAMINATION

SEMESTER / SESSION : SEM II / 2016/2017  
 COURSE NAME : ENGINEERING TECHNOLOGY MATERIALS

PROGRAMME CODE: 1 BDC/ 1 BDM  
 COURSE CODE : BDU 10603

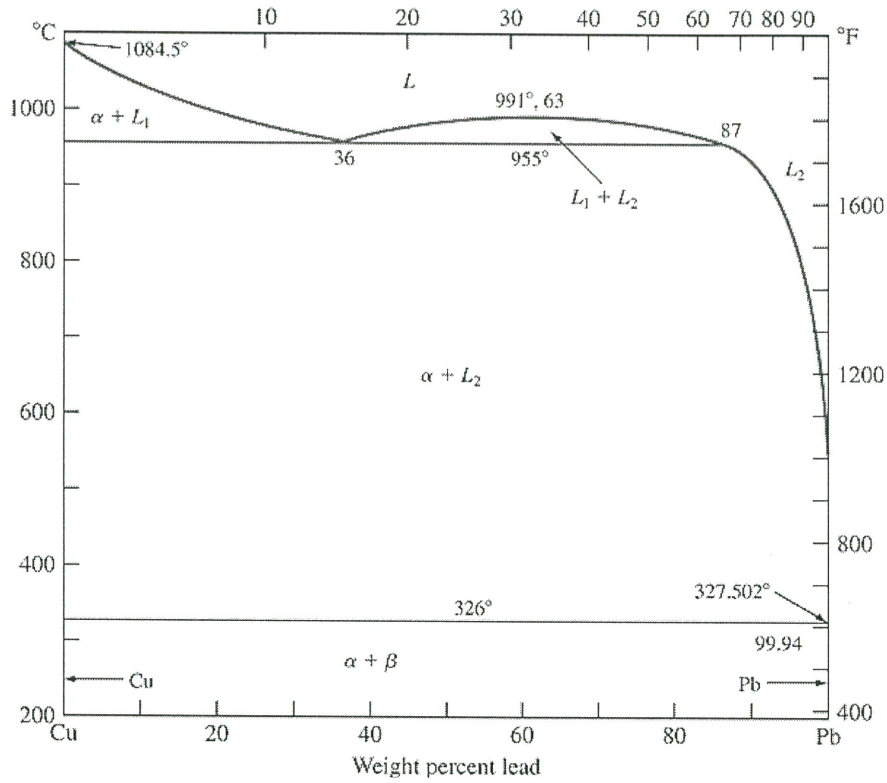


Figure Q4 (a): The Copper-Lead phase diagram.

UNIVERSITI TEKNOLOGI  
 MALAYSIA  
 INSTITUT TEKNOLOGI  
 MALAYSIA  
 INSTITUT TEKNOLOGI  
 MALAYSIA

TERBUKA