

## UNIVERSITI TUN HUSSEIN ONN MALAYSIA

## FINAL EXAMINATION **SEMESTER I SESSION 2015/2016**

**COURSE NAME** 

INTRODUCTION TO CHEMICAL

**ENGINEERING TECHNOLOGY** 

COURSE CODE : BNQ 10103

PROGRAMME

1 BNN :

:

**TEST DATE** 

DECEMBER 2015/ JANUARY 2016

**DURATION** 

2 HOURS :

INSTRUCTION : ANSWER ALL QUESTIONS

THIS QUESTION PAPER CONSISTS OF FOUR (4) PAGES

## CONFIDENTIAL

**BNO 10103** 

Q1	"Chemical engineer technologist" may be called "universal engineers" because their scientific
	and technical mastery is so extensive.

(a) Describe the term *chemical engineering technology*.

(3 marks)

(b) Demonstrate the jobs opportunity for chemical engineering technologist.

(5 marks)

(c) Assess FOUR (4) importance of units and conversions.

(4 marks)

(d) Show **THREE** (3) examples of derived dimension.

(3 marks)

(e) Analyze which of these equations are dimensionally homogenous:

i. 
$$x(m) = x_0(m) + 0.304(m/ft)v(ft/s)t(s) - 5m^2 + 0.5a(m/s^2)[t(s)]^2$$

ii.  $P(kg/ms^2) = 10135(Pa/atm) 1(kg/ms^2/Pa)P_o(atm) + p(kg/m^3)v(m/s)$ 

(4 marks)

(f) Define and differentiate empirical formula and molecular formula. Provide an example for each formula.

(6 marks)

Q2 (a) Calculate the mass of zinc in a 50.00 g sample of zinc nitrate, Zn(NO<sub>3</sub>).

(3 marks)

(b) 0.039 mol of calcium carbonate (CaCO<sub>3</sub>) is required in an experiment. Calculate mass of CaCO<sub>3</sub> needs to be weighed out.

(3 marks)

(c) A chemist needs 58.75 grams of urea, calculate how many grams of ammonia are needed to produce this amount?

 $2NH_3(g)+CO_2(g) \rightarrow (NH_2)_2CO(aq)+H_2O(1)$ 

(3 marks)

(d) List the law of conservation of mass in material balances.

(3 marks)

(e) List the general balance equation of material balances.

(2 marks)

(f) **Figure Q2** describes a distillation column problem for a mixture containing Y and Z. Assess and fined all flows and compositions.

(11 marks)

Q3	Process flow sheets are compact and precise diagrams that present a large amount of technical
	information about chemical processes.

(a) Differentiate **THREE** (3) major types of chemical process flow sheets.

(6 marks)

(b) Differentiate between batch, semi-batch and continuous process.

(6 marks)

(c) Describe the term *Biotechnology* 

(2 marks)

- (d) Describe the applications of biotechnology in the following areas:
  - i. Agricultural industry
  - ii. Bioprosessing

(6 marks)

(e) Explain the example of bioproduct developments & productions

(5 marks)

- Q4 Safety or loss prevention is the prevention of accidents through the use of appropriate technologies to identify the hazard of a chemical plant and eliminate them before an accident occurs.
  - (a) Identify **FOUR** (4) objectives of industrial safety and health.

(4 marks)

(b) Sketch **THREE** (3) symbols indicating different classes of hazardous substances.

(6 marks)

- (c) Accident and loss statistic are important measures of the effectiveness of safety programs.
  - (i) Describe **TWO** (2) methods of a safety measurement.

(4 marks)

(ii) A process has reported FAR of 3. If an employee works standard 8 hr shift 300 days per year, calculate the deaths per person per year.

(2 marks)

(iii) Demonstrate TWO (2) effects of prolong chemical exposure.

(4 marks)

(d) The dose-response curve describes the relationship between degree of exposure to a chemical (dose) and the magnitude of the effect (response) in the exposed organism. Illustrate and interpret the dose-response curve.

(5 marks)

END OF QUESTION -

COURSE NAME



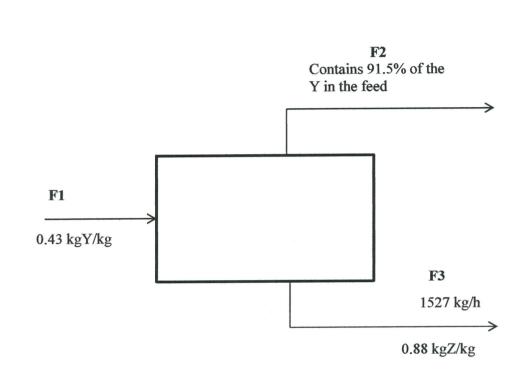


FIGURE Q2