

UNIVERSITI TUN HUSSEIN ONN MALAYSIA

FINAL EXAMINATION SEMESTER I **SESSION 2015/2016**

COURSE NAME

SEMICONDUCTOR MATERIAL

ANALYSIS

COURSE CODE

: BED 40702

PROGRAMME

BACHELOR OF ELECTRONIC

ENGINEERING WITH HONOURS

EXAMINATION DATE : DECEMBER 2015 / JANUARY 2016

DURATION

: 2 HOURS

INSTRUCTION

: ANSWER ALL QUESTIONS

THIS QUESTION PAPER CONSISTS OF FOUR (4) PAGES

Q1	(a)	This question is based on Figure Q1. (i) Identify TWO (2) types of chemical bonding.	
		(i) Identity I WO (2) types of ellement boliding.	
		(ii) By the aid of Lewis diagram, give TWO (2) examples of different che	emical
		bonding. (4 n	narks)
	(b)	(i) Define the intrinsic and extrinsic semiconductors. (2 n	narks)
		(ii) Show a schematic diagram of silicon crystal lattice if this material is with a dopant to form <i>p</i> -type semiconductor. (4 n	added marks)
	(c)	(i) Define the generation and recombination process of a semiconductor. (2)	marks)
		(ii) Show a schematic diagram of energy level for recombination process. (6)	marks)
Q2	(a)	Arrange an energy band for a <i>p</i> -type Si doped with 1 ppm Boron. (10 s	marks)
	(b)	(i) Explain how an addition of Phosphorus helps to increase the conductive silicon.	vity of marks)
		(ii) Organize a typical structure of electron movement in silicon crystal when this material is doped with Phosphorus. (6 1)	lattice
Q3	(a)	(i) Illustrate the structural shape of Carbon-Silicon. (3 r	marks)
		(ii) Named the structural shape of Carbon-Silicon. (1	mark)
		(iii) List SIX (6) key properties of Carbon-Silicon compound (6 r	marks)
	(b)	Using a table, demonstrate THREE (3) key properties of Silicon-Germa Silicon-Tin and Germanium-Tin.	anium, marks)

Q4 (a) Using a table, distinguish the thermal conductivity, band gap and application of Boron Nitride, Aluminium Nitride and Gallium Nitride.

(10 marks)

(b) By the aid of diagram, discuss the direct and indirect band gap.

(10 marks)

Q5 (a) Using a table,

i) Outline **SEVEN** (7) compounds from binary II-VI.

(7 marks)

(ii) Indicate the band gap energy of each outlined compound.

(7 marks)

(b) Choose one compound from binary II-VI and analyze the selected compound.

(6 marks)

- END OF QUESTION -

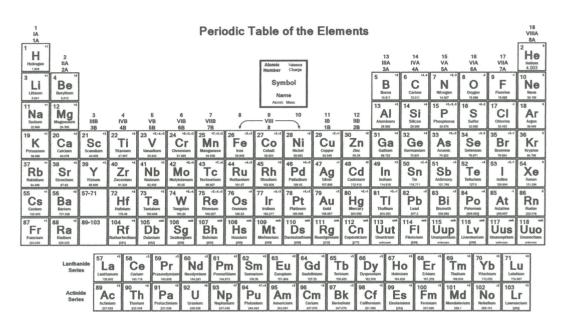
FINAL EXAMINATION

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FIGURES Q1